

Analysis of Ni-Zn Batteries Performance for Hybrid Light-Vehicles Applications

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Outline of the presentation

- Introduction
- Ni-Zn batteries short-term characterization and comparison with lead-acid ones
- Ni-Zn batteries long-term characterization
- Ni-Zn batteries model for capacity estimation
- Conclusions

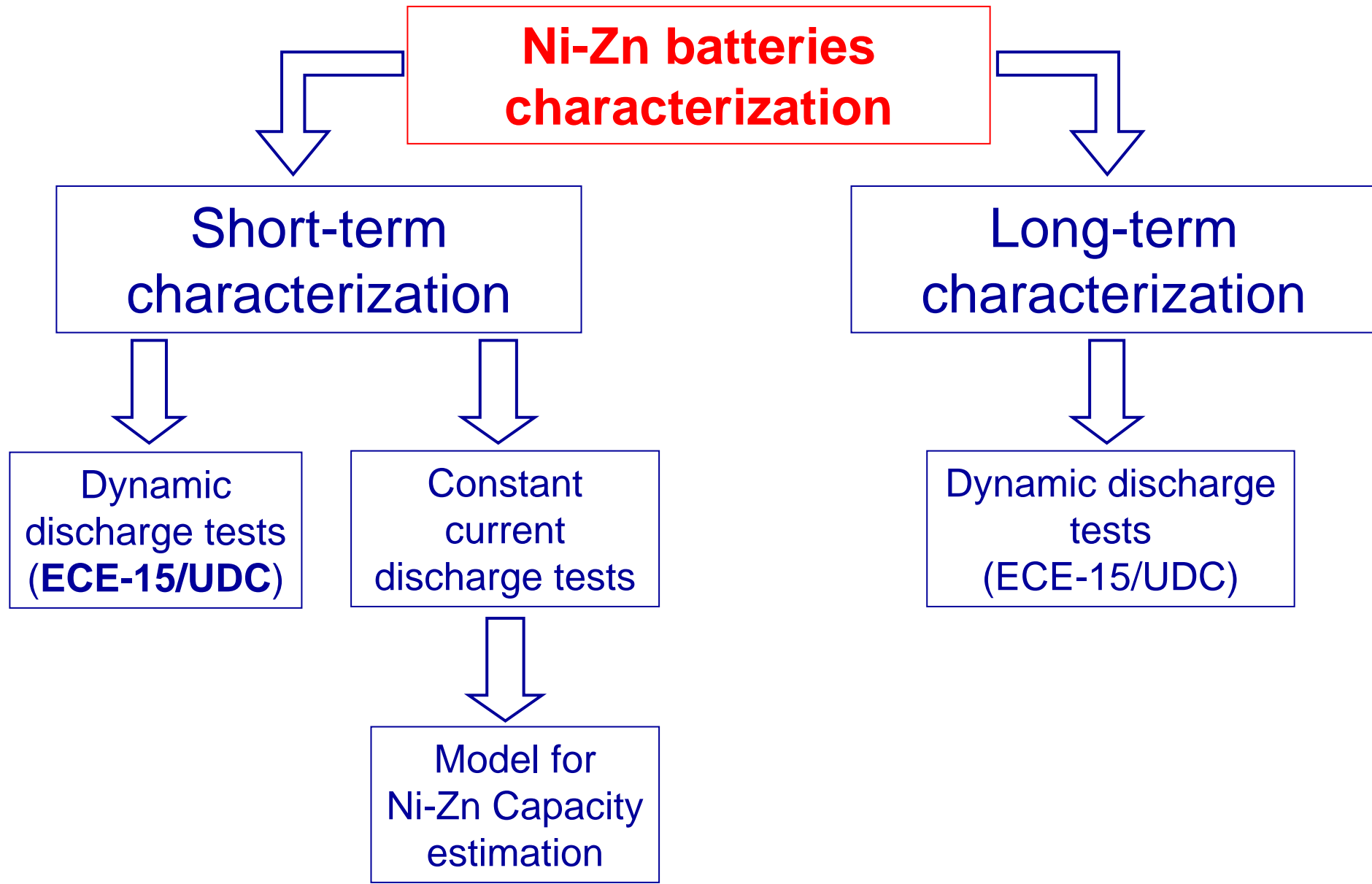
Introduction

Improvement of the performance of hybrid-electric vehicles

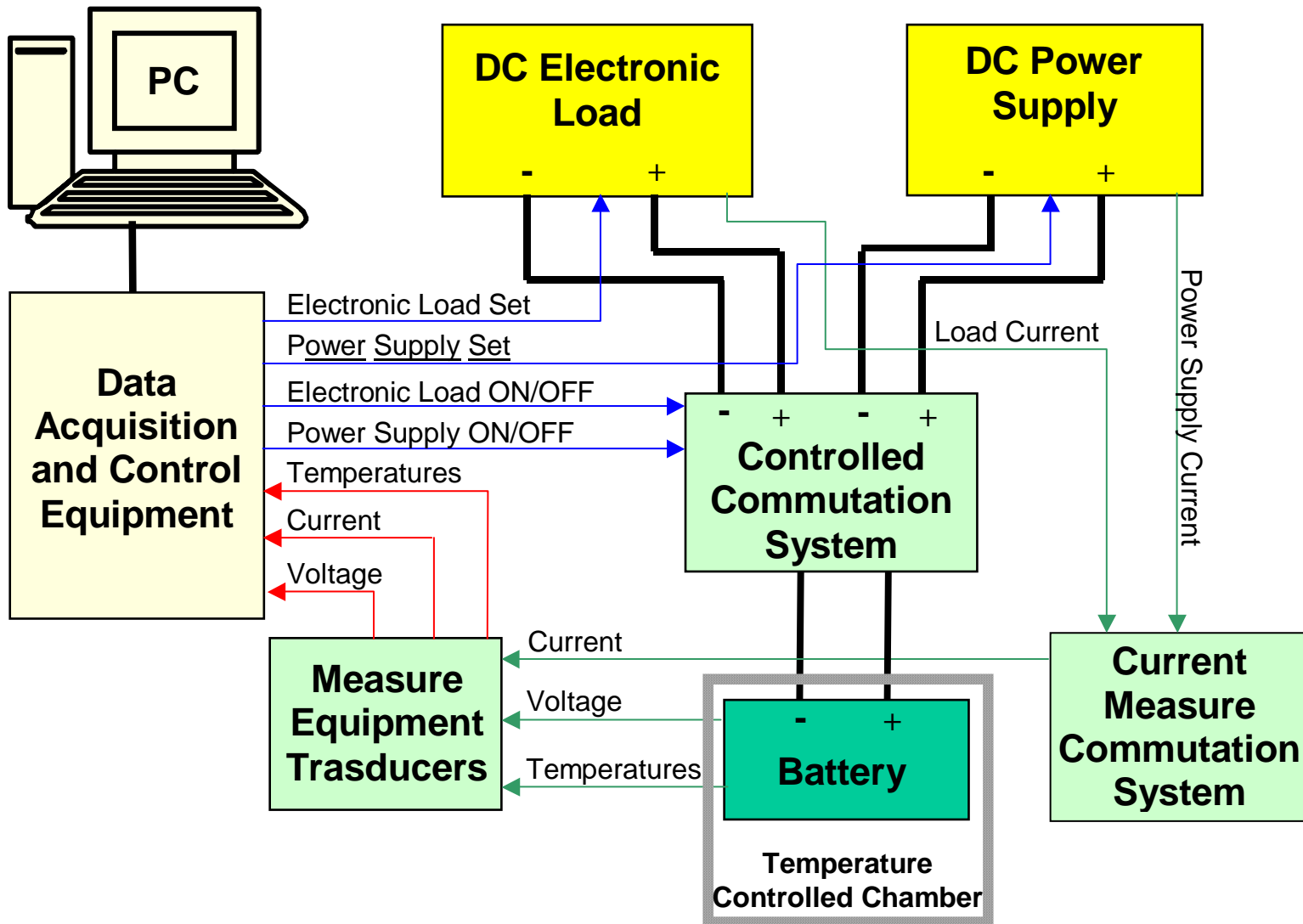


Progresses in the technology of the electrical energy storage

Within the *electrochemical systems for the electrical energy storage*, recent improvements in the reliability of *Ni-Zn* batteries make this technology one of the most promising for hybrid-electric vehicles applications.



PC-controlled bench



Ni-Zn batteries short-term characterization and comparison with lead-acid ones



Ni-Zn battery characteristics:

- Evercel; model: Ni-Zn 40-12;
 - **38.5 Ah at C/20;**
 - **32.7 Ah at C/1;**
 - **31.1 Ah at 3C;**
 - **20°C;**

- weight **7.88 kg.**



Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*



Lead-acid battery characteristics:

- Hawker; model: Genesis G12V26Ah10EP;
 - **26 Ah at C/10, 22.5 at C/5;**
 - **20°C;**

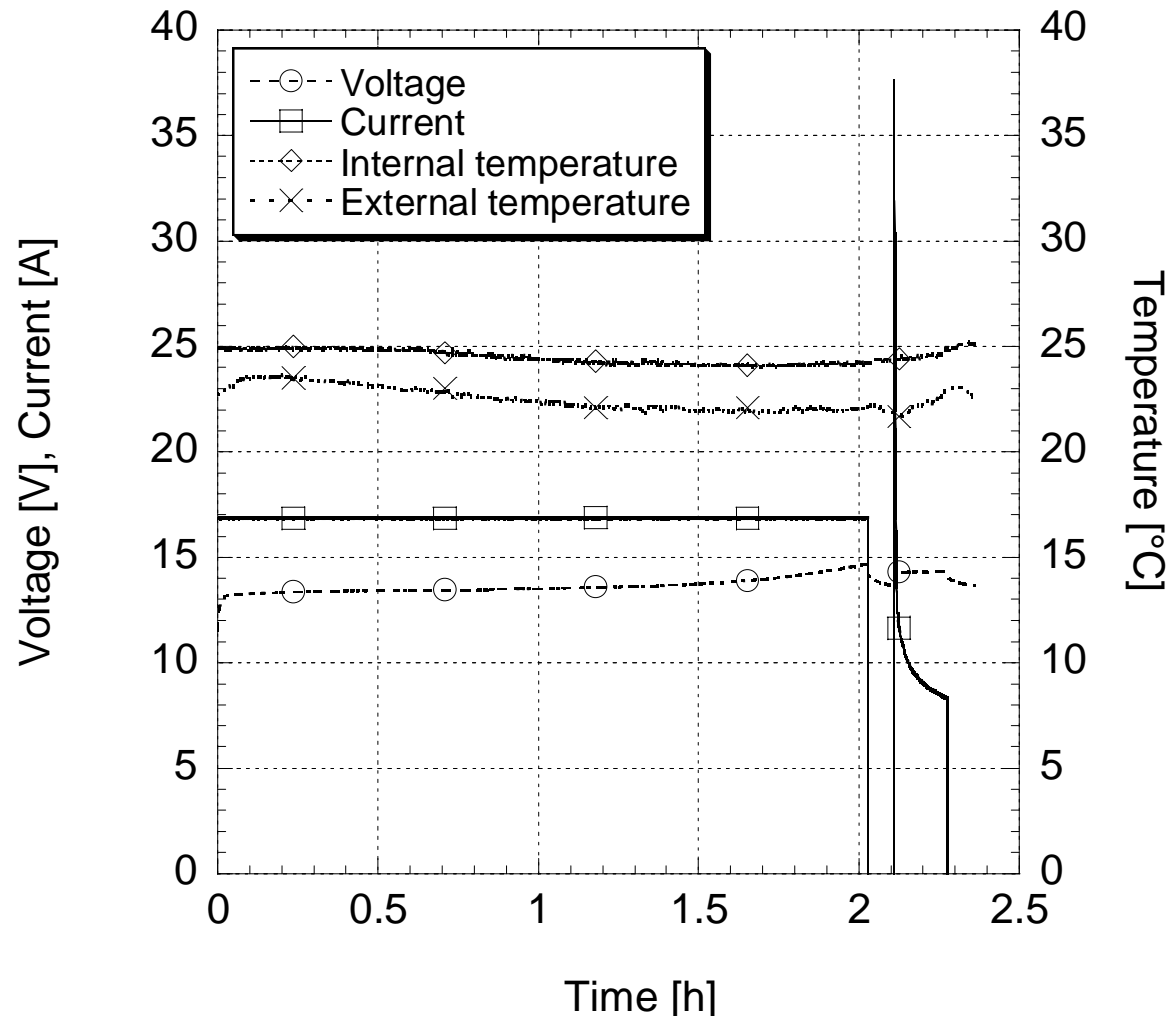
- weight **10.1 kg.**



Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*

Ni-Zn Charge

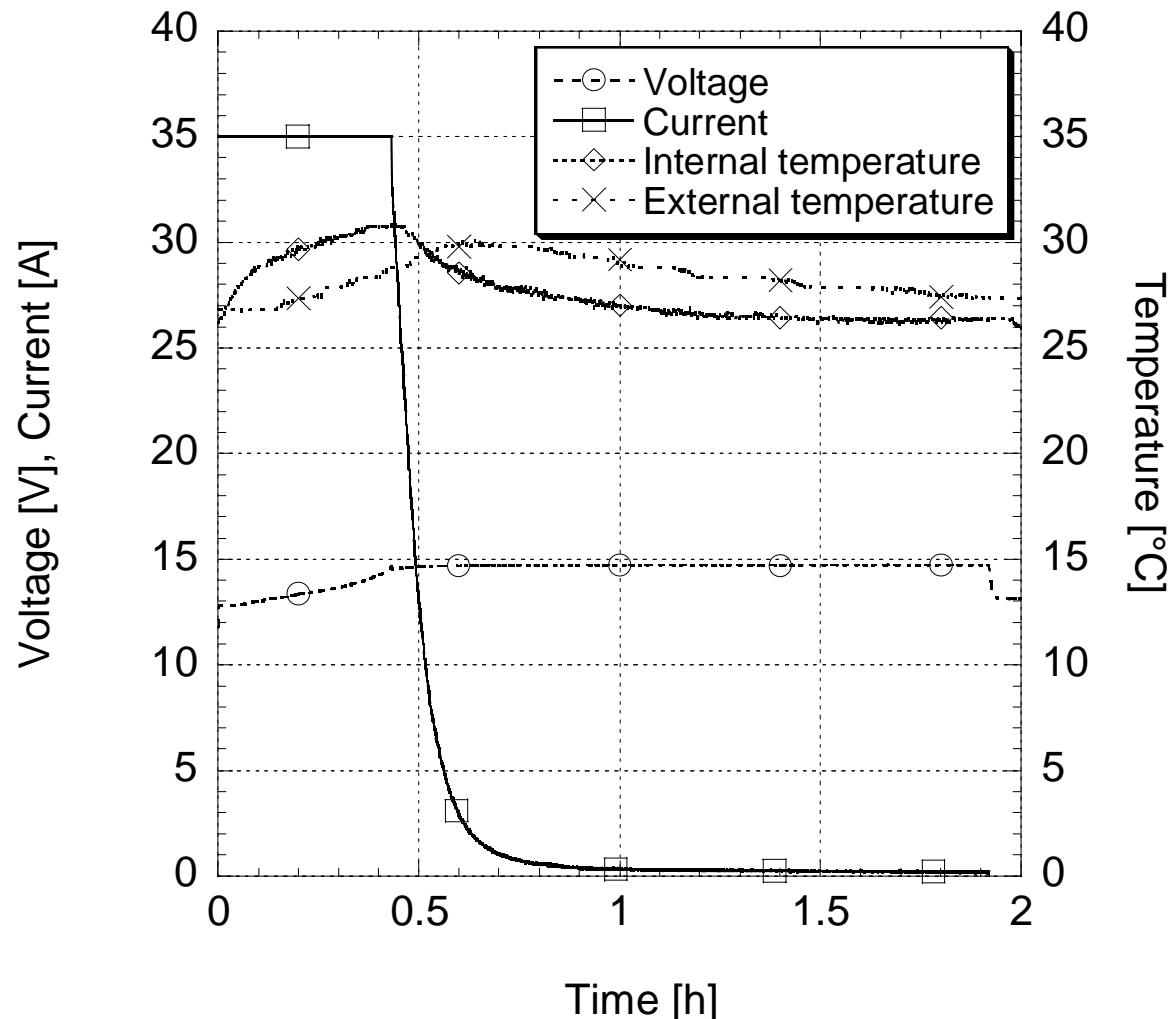
- phase 1: charge at constant current $C/1.75$, maximum voltage: 14.7 V;
- phase 2: stop for 5 minutes;
- phase 3: charge at constant voltage 14.35 V, minimum current $C/4$.



Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*

Lead-acid Charge

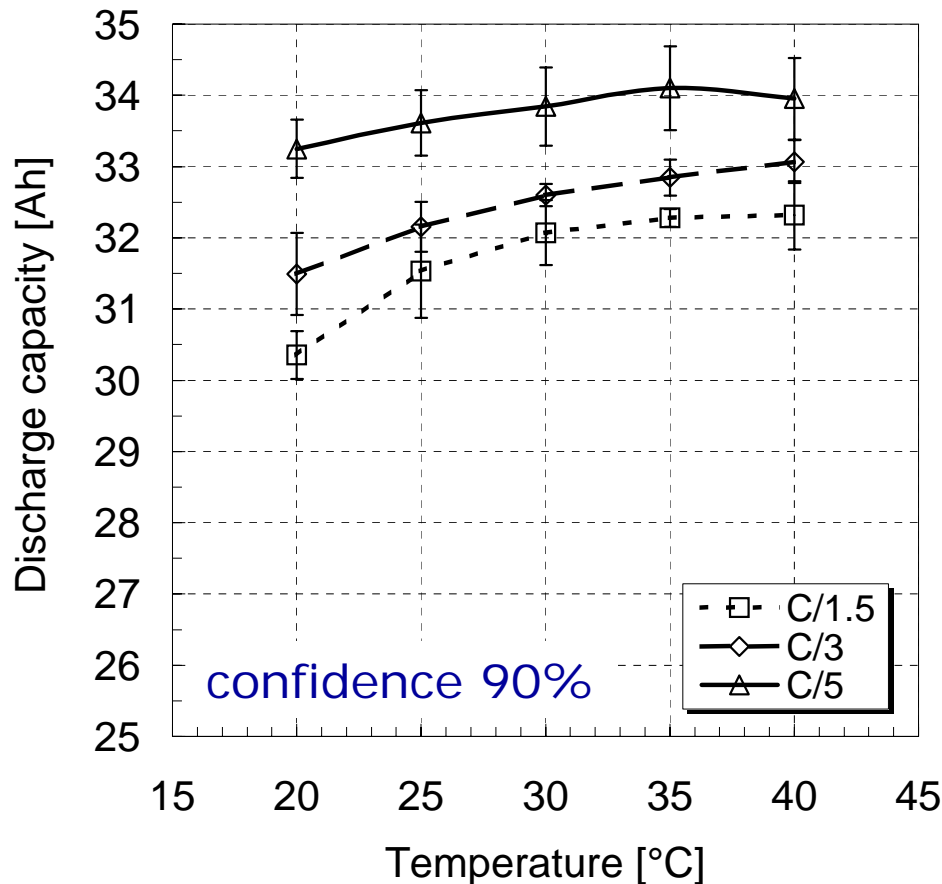
- phase 1: charge with constant current equal to $0.4 \times C/10$ for a charging voltage less than 14.7 V (the duration of the phase 1 is T1);
- phase 2: charge with constant voltage equal to 14.7 V for time equal to $T2 = 2h - T1$.



Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*

Ni-Zn discharges with constant current

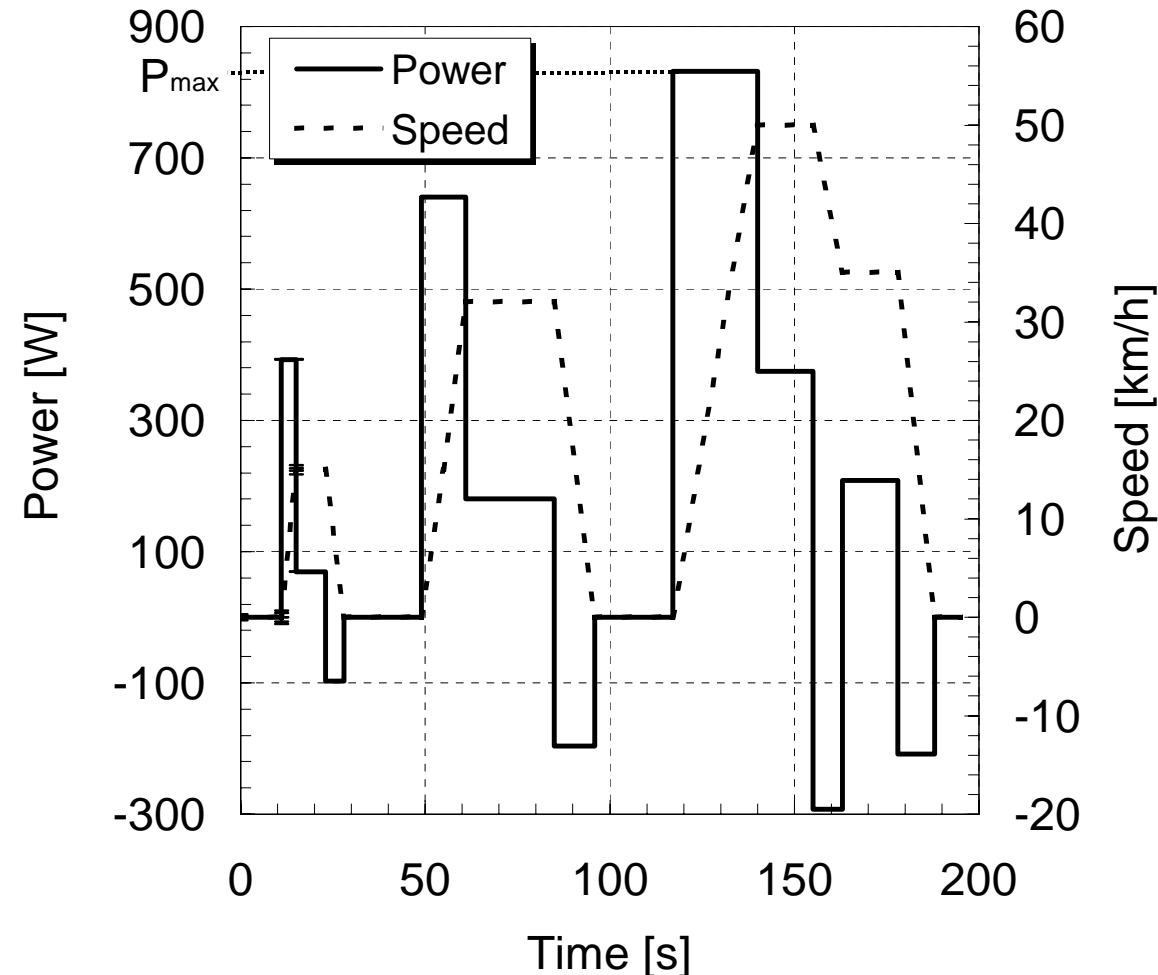
Ni-Zn battery performances as function of *discharge rate* and *temperature*.



The decrease of the capacity for temperatures above 30° is probably due to the Nickel electrodes that show a kind of unstable behavior above this temperature.

Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*

Dynamic discharges based on the urban part of the **ECE-15** cycle: **ECE-15/UDC**

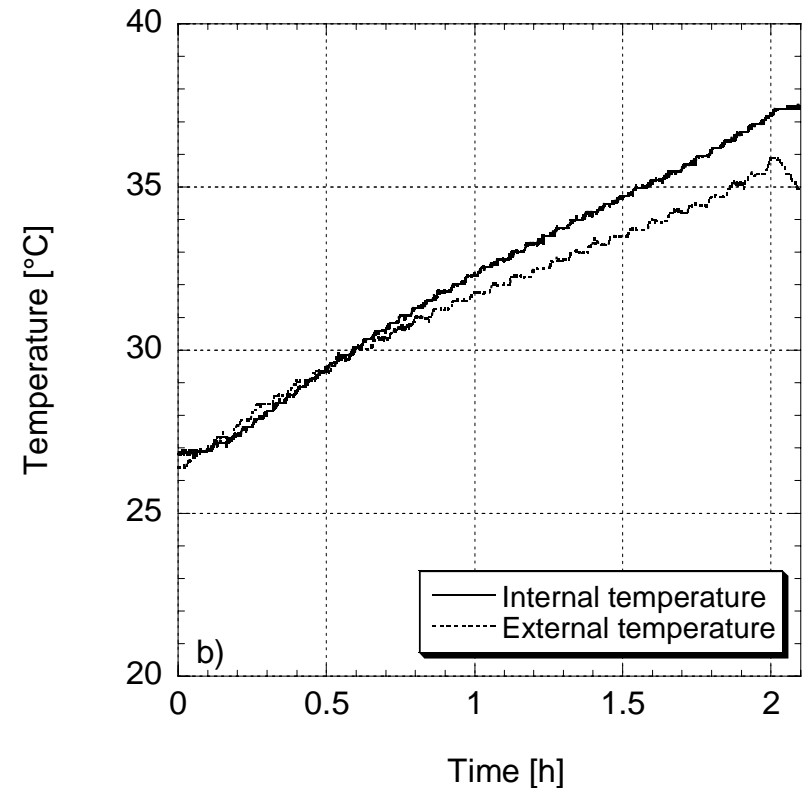
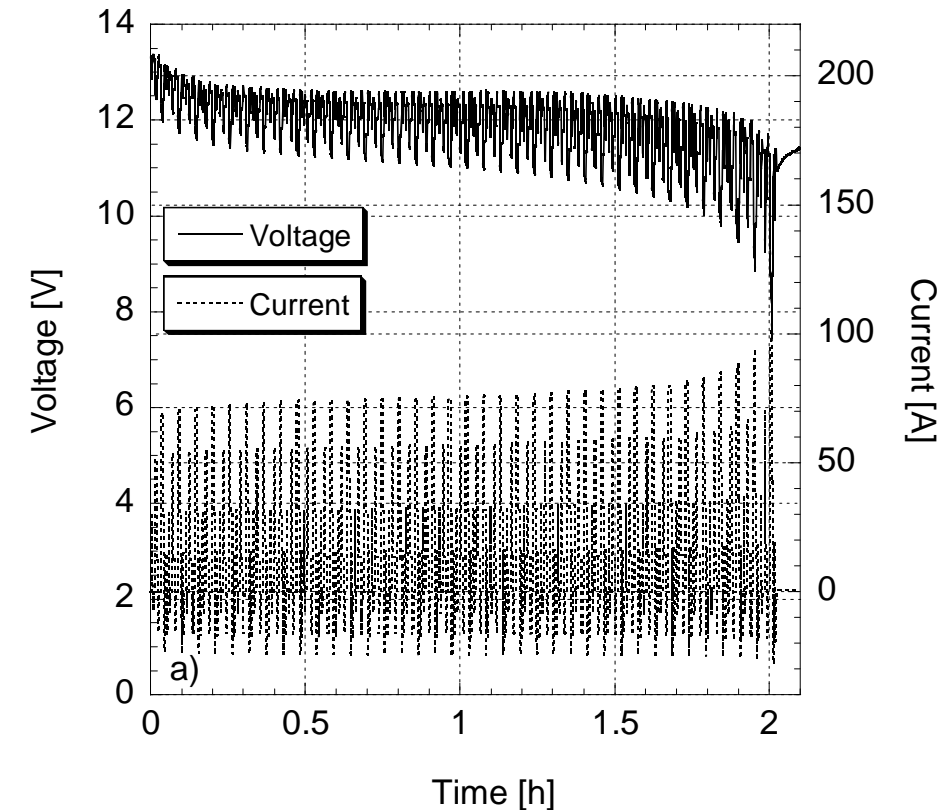


This type of tests is suitable for the simulation of the batteries performance when used on electrical vehicles and, in particular, when urban paths are of interest.

Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*

Ni-Zn dynamic discharges based on the **ECE-15/UDC** mini-cycle.

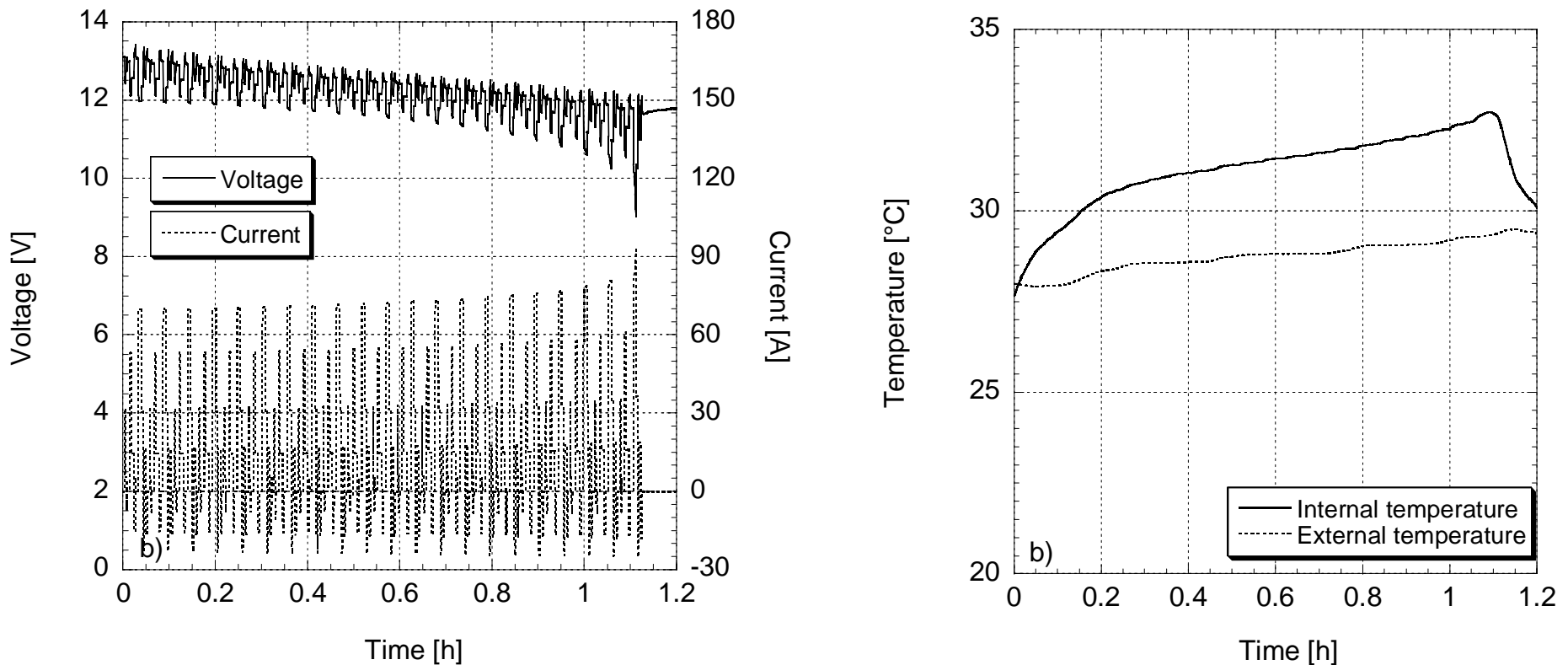
(Figure show recorded quantities for the Ni-Zn elements under test. The ambient temperature is of 25 °C)



Ni-Zn batteries short-term characterization and comparison with lead-acid ones *Cont.*

Lead-acid dynamic discharges based on the **ECE-15/UDC** mini-cycle for the lead-acid elements.

(Figure show recorded quantities for the lead acid elements under test. The ambient temperature is of 25 °C)



Ni-Zn batteries short-term characterization and comparison with lead-acid ones

Cont.

Comparison between Ni-Zn and lead-acid performances.

	Ni-Zn specific energy [Wh/kg]	Lead-acid specific energy [Wh/kg]
C/5 constant current discharge	49.6	40.7
ECE-15/UDC dynamic discharge	27.9	20.3

Note: average values obtained from 6 elements for Ni-Zn batteries and 2 elements for lead-acid ones

Ni-Zn batteries long-term characterization

A long-term characterization has been performed considering the *on-board vehicle use* of the Ni-Zn batteries. Hence the urban part of the ECE-15 cycle has been adopted in order to reproduce the power absorption conditions met on a light electrical vehicle.

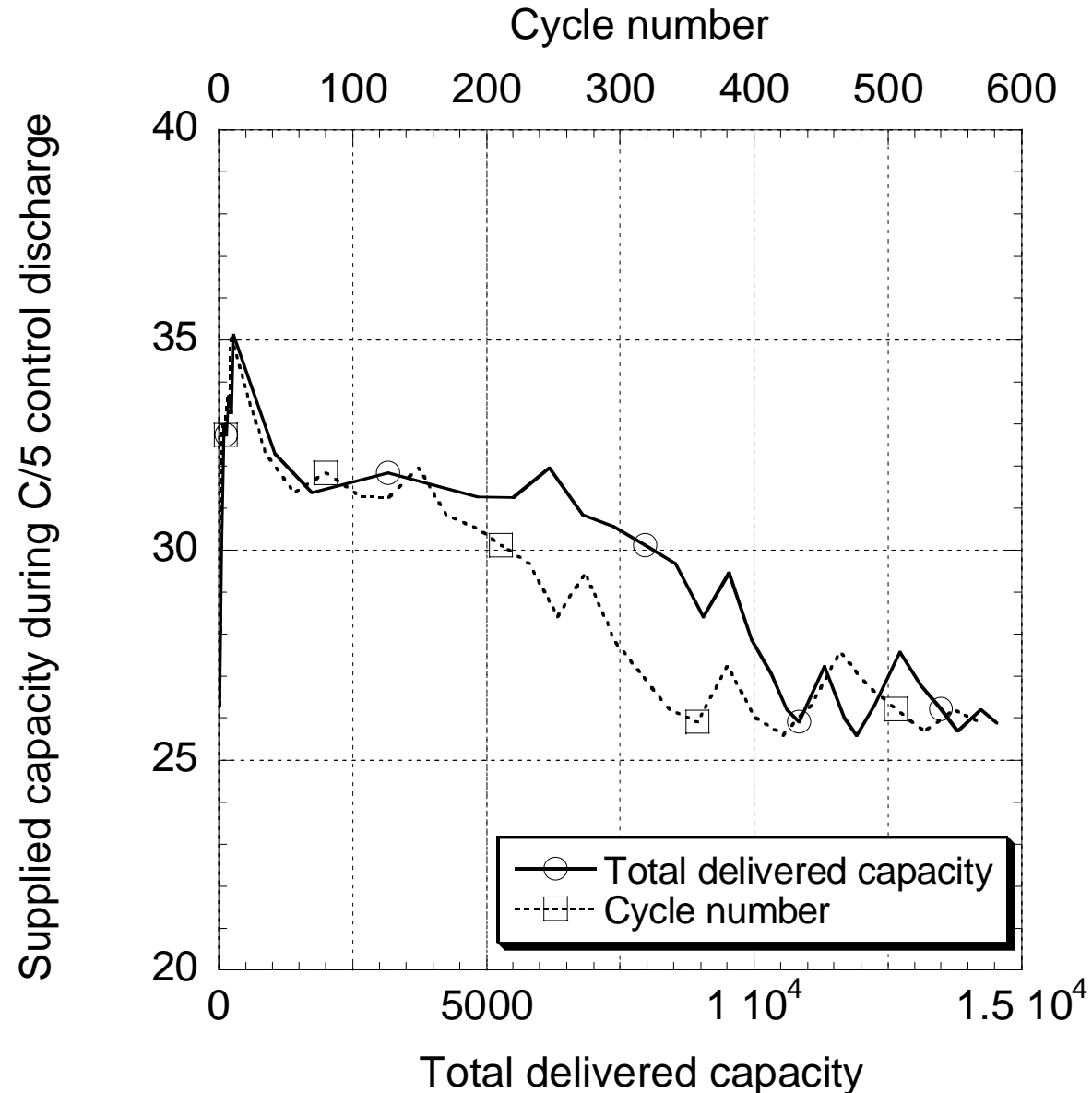
Long-term test sequences:

- ❑ series of 20 charge – discharges macro cycles;
- ❑ C/5 control discharge to evaluate the effective battery capacity and, in turns, the battery end-of-life.

Ni-Zn batteries long-term characterization

Cont.

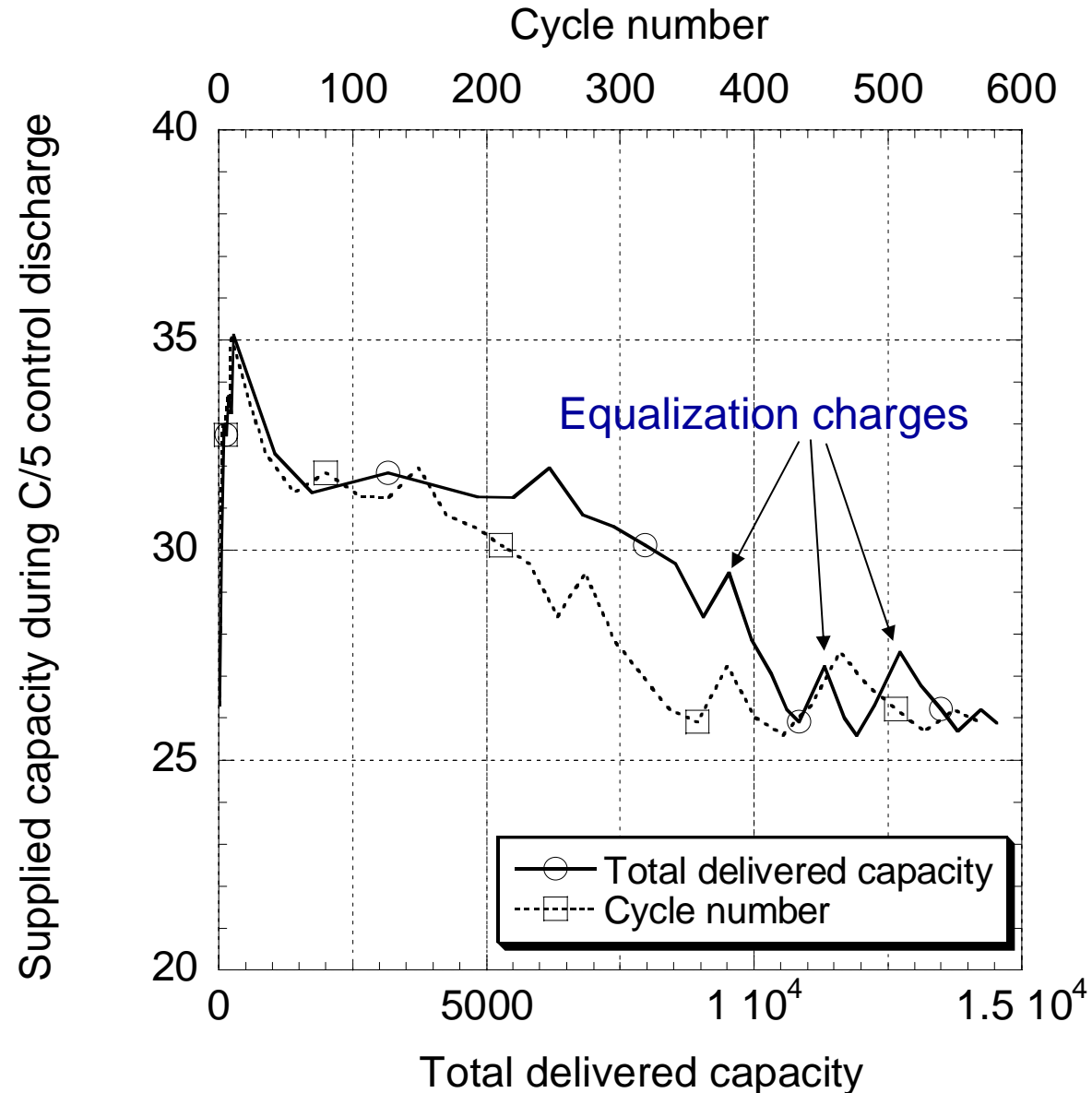
Capacity of the Ni-Zn battery delivered during the control discharges as a function of the total delivered capacity



Ni-Zn batteries long-term characterization

Cont.

Capacity of the Ni-Zn battery delivered during the control discharges as a function of the total delivered capacity



Ni-Zn batteries long-term characterization

Cont.

The measured value of the battery capacity during C/5 control discharge, is compared with the rated battery capacity at C/5 discharge rate and, when a decrease of 20% the rated capacity is reached, the battery is considered at the end of its life.

Note the severity of such a criterion, since for hybrid vehicle applications the capacity needed for the vehicle to cover a standard one-day route is reasonably much lower than 80% the rated one.

The long term tests show a performances of the Ni-Zn batteries of *600 total cycles* with an approximately total delivered capacity of *15000 Ah*.

Ni-Zn model for capacity estimation

The proposed model for the Ni-Zn capacity estimation, is inspired by a previous one, presented in [1], which was conceived to predict the behavior of lead-acid batteries according to the following relation:

$$C(I, \theta) = \frac{Kc \cdot C_0^* \cdot \left(1 + \frac{\theta}{\theta_f}\right)^\varepsilon}{1 + (Kc - 1) \cdot (I/I^*)^\delta}$$

- θ is the electrolyte temperature;
- I is the battery current during a constant discharge profile;
- θ_f is the electrolyte solidification temperature;
- $Kc, C_0^*, \varepsilon, \delta, I^*$ are model parameters to be identified by means of tests described in [1].

Such a model however, if applied to Ni-Zn batteries has been found to provide only a moderate agreement between experimental results and predicted ones.

Possible reasons:

1. capacity variation due to temperature and due to the discharge current are 'decoupled'. Assumption reasonable for lead-acid batteries (temperature coefficient);
2. Nickel electrodes show a kind of unstable behavior above 30°C

Proposed model:

$$C(I, \theta) = \frac{C_n \cdot [I_n / I + (\theta - \theta_n)]^\varepsilon}{(I / I^*)^\delta}$$

- θ is the electrolyte temperature;
- I is the battery current during a constant discharge profile;
- C_n is the rated capacity at current I_n and temperature θ_n ;
- ε, δ, I^* are model parameters to be identified by means of short-time tests.

Identification of the parameters ε , δ , I^*

An estimation of the three above parameters can be achieved by carrying out more than 3 measurements and by using them, together with a non-linear least squares method, to infer ε , δ and I^* .

The values of parameters ε , δ and I^* so determined are:

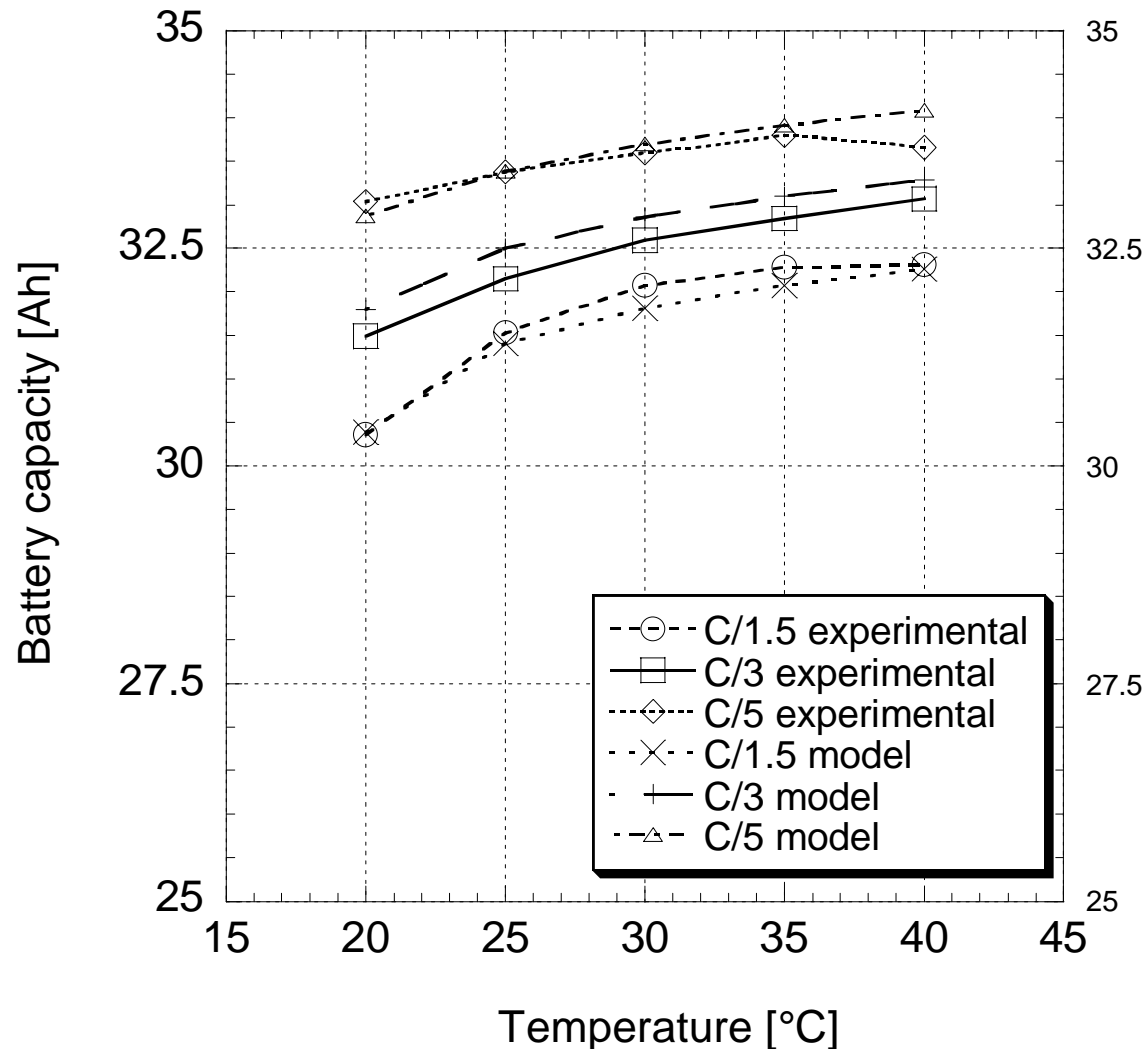
$$\varepsilon = 0.0225$$

$$\delta = 0.0428$$

$$I^* = 21.8 \text{ A}$$

with a residual norm of 2.7%

Comparison between measurements and model predicted capacities at different temperatures and discharge rates.



Conclusions

The comparison between the two battery technologies under test has shown that the Ni-Zn one represents a promising solution for the hybrid/electrical vehicles application.

In particular the short term performance analysis shows the superior behavior of the Ni-Zn technology, as compared with the lead-acid one, when dynamic discharge profiles are considered. In addition, more homogeneous values of the supplied capacity as a function of the type of discharge, discharge rate and temperature, has been found for the Ni-Zn batteries.

Concerning the long-term performances, interesting values of total supplied capacity and total number of cycles are obtained for the Ni-Zn batteries under test. This result is particular interesting considering the type of selected discharges based on the ECE-15/UDC dynamic cycle.

An engineering model aimed at estimating the capacity supplied by the Ni-Zn batteries, as a function of the constant discharge current and temperature, has been proposed and compared, with sufficient agreement, with experimental results.